

# Cambridge International AS & A Level

CHEMISTRY 9701/36

Paper 3 Advanced Practical Skills 2

October/November 2024

CONFIDENTIAL INSTRUCTIONS

This document gives details of how to prepare for and administer the practical exam.

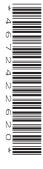
The information in this document and the identity of any materials supplied by Cambridge International are confidential and must NOT reach candidates either directly or indirectly.

The supervisor must complete the report at the end of this document and return it with the scripts.

#### **INSTRUCTIONS**

 If you have any queries regarding these confidential instructions, contact Cambridge International stating the centre number, the syllabus and component number and the nature of the query.
 email info@cambridgeinternational.org

phone +44 1223 553554



# General information about practical exams

Centres must follow the guidance on science practical exams given in the Cambridge Handbook.

# Safety

Supervisors must follow national and local regulations relating to safety and first aid.

Only those procedures described in the question paper should be attempted.

Supervisors must inform candidates that materials and apparatus used in the exam should be treated with caution. Suitable eye protection should be used where necessary.

The following hazard codes are used in these confidential instructions, where relevant:

CcorrosiveMHmoderate hazardHHhealth hazardTacutely toxicFflammableOoxidising

**N** hazardous to the aquatic environment

Hazard data sheets relating to substances used in this exam should be available from your chemical supplier.

#### Before the exam

- The packets containing the question papers must **not** be opened before the exam.
- It is assumed that standard school laboratory facilities, as indicated in the *Guide to Planning Practical Science*, will be available.
- Spare materials and apparatus for the tasks set must be available for candidates, if required.

### **During the exam**

- It must be made clear to candidates at the start of the exam that they may request spare materials and apparatus for the tasks set.
- Where specified, the supervisor must perform the experiments and record the results as instructed.
   This must be done out of sight of the candidates, using the same materials and apparatus as the candidates.
- Any assistance provided to candidates must be recorded in the supervisor's report.
- If any materials or apparatus need to be replaced, for example, in the event of breakage or loss, this must be recorded in the supervisor's report.

# After the exam

- The supervisor must complete a report for each practical session held and each laboratory used.
- Each packet of scripts returned to Cambridge International must contain the following items:
  - the scripts of the candidates specified on the bar code label provided
  - the supervisor's results relevant to these candidates
  - the supervisor's reports relevant to these candidates
  - seating plans for each practical session, referring to each candidate by candidate number
  - the attendance register.

# Specific information for this practical exam

During the exam, the supervisor (**not** the invigilator) must do all the experiments and record the results on a spare copy of the question paper, clearly labelled 'supervisor's results'.

If chemicals are prepared in more than one batch, clearly labelled supervisor's results must be provided for each batch. The candidates using each batch must be listed on the supervisor's report.

# **Apparatus**

- $1 \times 50 \, \text{cm}^3$  measuring cylinder
- 1 x 250 cm<sup>3</sup> plastic or glass measuring cylinder
- $1 \times 100 \, \text{cm}^3 \, \text{beaker}$
- 2 × stand and clamp
- $1 \times \text{tub}$  suitable for acting as a trough (minimum capacity  $1 \text{ dm}^3$ )
- $1 \times \text{side-arm}$  conical flask with bung connected to approximately 50 cm of plastic/rubber delivery tube OR  $150 \, \text{cm}^3$  or  $250 \, \text{cm}^3$  conical flask with one-hole bung connected to approximately  $50 \, \text{cm}$  of plastic/rubber delivery tube. Label this flask **X**.
- 1 × glass rod
- 1 × stop-clock to measure to an accuracy of 1 second
- 2 × teat/dropping pipette
- 1 × spatula
- $1 \times \text{crucible with lid (approximate capacity } 15 \text{ cm}^3)$
- 1 × crucible tongs
- 1 × pipe-clay triangle
- 1 × tripod
- 1 × Bunsen burner
- 1 × heat-proof mat
- 1 × test-tube holder
- 2 × boiling tube
- 1 × hard-glass test-tube
- 6 × test-tube\*
- 1 × test-tube rack

balance, single-pan, direct reading, minimum accuracy 0.01g (1 per 8-12 candidates) weighing to 200g

1 × wash bottle containing distilled water

paper towels

1 × pen for labelling glassware

red and blue litmus papers

aluminium foil

wooden splints

the apparatus normally used in the centre for use with limewater in testing for carbon dioxide

\*Candidates are expected to rinse and reuse test-tubes where possible. Additional tubes should be available.

the question paper that FB 1 is phosphoric acid.

| label                          | per<br>candidate   | identity   | notes  |
|--------------------------------|--------------------|--|--|
| FB 1 [MH]                      | 1.25g              | mixture of anhydrous sodium carbonate and sodium chloride in an approximately 2:1 ratio by mass        | 1.25 $\pm$ 0.05g mixture containing Na <sub>2</sub> CO <sub>3</sub> <b>[MH]</b> and NaC <i>l</i> in an approximate ratio of 2:1 by mass. The solids should be well mixed and supplied in a stoppered container.  |
| FB 2                           | 70 cm <sup>3</sup> | 0.60 moldm <sup>-3</sup> hydrochloric acid   | Dilute $300\mathrm{cm}^3$ of $2.00\mathrm{moldm}^{-3}$ HC $l$ to $1.0\mathrm{dm}^3$ . For preparation of $2.00\mathrm{moldm}^{-3}$ HC $l$ see current syllabus.  |
| FB 3 [MH]                      | 2.0g               | mixture of magnesium carbonate<br>and anhydrous sodium carbonate in<br>approximately 2:1 ratio by mass | $2.0 \pm 0.1$ g mixture containing MgCO <sub>3</sub> and Na <sub>2</sub> CO <sub>3</sub> <b>[MH]</b> in the ratio 2:1 by mass. The solids should be well mixed and supplied in a stoppered container. Any form of (basic) magnesium carbonate is suitable. |
| FB 4 [C][MH][N]                | 59                 | mixture of approximately equal masses of ammonium chloride and hydrated zinc sulfate                   | $5.0 \pm 0.1$ g mixture containing NH <sub>4</sub> C <i>l</i> [MH] and ZnSO <sub>4</sub> •7H <sub>2</sub> O [C][MH][N] in approximate ratio 1:1 by mass. The solids should be well mixed and supplied in a stoppered container.                            |
| aqueous<br>sodium<br>carbonate | 5 cm <sup>3</sup>  | 0.50 moldm <sup>-3</sup> sodium carbonate  | Dissolve 53.0g Na <sub>2</sub> CO <sub>3</sub> <b>[MH]</b> or 143.0g Na <sub>2</sub> CO <sub>3</sub> •10H <sub>2</sub> O <b>[MH]</b> in each dm <sup>3</sup> of solution.  |
| universal<br>indicator [F]     | 2cm <sup>3</sup>   | universal indicator paper or solution  | Full range UI paper or solution is suitable.   |

| © CI                         | per                | identity   | notes   |
|------------------------------|--------------------|--|---|
|                              | candidate          |  |   |
| dilute hydrochloric acid     | 10 cm <sup>3</sup> | 2.0 mol dm <sup>-3</sup> HCl                                     |   |
| dilute nitric acid [C]       | 10 cm <sup>3</sup> | 2.0 mol dm <sup>-3</sup> HNO <sub>3</sub>                        | See preparation instructions in the current   |
| dilute sulfuric acid [MH]    | 10 cm <sup>3</sup> | 1.0 mol dm <sup>-3</sup> H <sub>2</sub> SO <sub>4</sub>          | syllabus.   |
| aqueous ammonia [C][MH][N]   | 10 cm <sup>3</sup> | 2.0 mol dm <sup>-3</sup> NH <sub>3</sub>                         | If necessary, each of these reagents can be   |
| aqueous sodium hydroxide [C] | 20 cm <sup>3</sup> | 2.0 mol dm <sup>-3</sup> NaOH                                    | provided as a communal supply for groups of up to   |
| aqueous barium chloride      | 10 cm <sup>3</sup> | 0.1 mol dm <sup>-3</sup> BaCl <sub>2</sub>                       | כמותימונים.   |
| or<br>aqueous barium nitrate |                    | or<br>0.1 mol dm <sup>-3</sup> Ba(NO <sub>3</sub> ) <sub>2</sub> | Invigilators must be alert to the risk of contamination and the opportunity for malpractice |
| limewater [MH]               | 10 cm <sup>3</sup> | saturated aqueous calcium hydroxide, Ca(OH) <sub>2</sub>         | when using a communal supply.   |
| aqueous silver nitrate       | 10 cm <sup>3</sup> | 0.05 moldm <sup>-3</sup> AgNO <sub>3</sub>                       |   |

An excess of at least 10% of each material must be prepared to cover accidental loss. aqueous silver nitrate
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All solutions must b

All solutions must be thoroughly mixed.

If you are unable to source any of these chemicals, you must contact Cambridge International as far as possible in advance of the exam for advice.

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# Supervisor's report

| Syllabus and component number |      |      | / |      |  |
|-------------------------------|------|------|---|------|--|
| Centre number                 |      |      |   |      |  |
| Centre name                   | <br> | <br> |   | <br> |  |
| Time of the practical session | <br> | <br> |   | <br> |  |
| Laboratory name/number        | <br> | <br> |   | <br> |  |

Give details of any difficulties experienced by the centre or by candidates (include the relevant candidate names and candidate numbers).

You must include:

- any difficulties experienced by the centre in the preparation of materials
- any difficulties experienced by candidates, e.g. due to faulty materials or apparatus
- any specific assistance given to candidates.

| If chemicals have been prepared in more than one batch, list the candidates using each bat | If ( | chemicals have be | een prep | pared in more | than on | e batch, | list the | candidates | using e | each b | ato |
|--|------|-------------------|----------|---------------|---------|----------|----------|------------|---------|--------|-----|
|--|------|-------------------|----------|---------------|---------|----------|----------|------------|---------|--------|-----|

# Declaration

- 1 Each packet that I am returning to Cambridge International contains all of the following items:
  - the scripts of the candidates specified on the bar code label provided
  - the supervisor's results relevant to these candidates
  - the supervisor's reports relevant to these candidates
  - seating plans for each practical session, referring to each candidate by candidate number
  - the attendance register.
- 2 Where the practical exam has taken place in more than one practical session, I have clearly labelled the supervisor's results, supervisor's reports and seating plans with the time and laboratory name/number for each practical session.
- 3 I have included details of difficulties relating to each practical session experienced by the centre or by candidates.
- 4 I have reported any other adverse circumstances affecting candidates, e.g. illness, bereavement or temporary injury, directly to Cambridge International on a *special consideration form*.

| Signed                   | (supervisor) |
|--------------------------|--------------|
| Name (in block capitals) |              |

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